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IMPORTANT SAFETY INFORMATION --PLEASE ROUTE TO COMPANY SAFETY OFFICER. DO NOT DISCARD.

PRODUCTS AND INFORMATION FOR THE LABORATORY.
CONTACT YOUR LOCAL FISHER BRANCH FOR FILMS, BRO-ISHER SCIENTIFIC HAS A OMPLETE LINE OF SAFETY CATALOGS AND PRO-

SYRACUSE CITY SCHOOL DIST 644 MADISON ST SYRACUSE NY 13210 13210

YOUR FISHER SALES REPRESENTATIVE OR YOUR IF NAME AND/OR ADDRESS

LOCAL FISHER BRANCH.

THIS PACKET MAY CONTAIN MSDS FOR PRODUCTS MAN-UFACTURED BY OTHERS AND DISTRIBUTED BY FISHER SCIENTIFIC COMPANY. THESE MSDS WERE PREPARED BY THE MAHUFACTURER AND FISHER DISCLAIMS ALL REQUIRED MATERIAL SAFETY-DATA SHEETS (MSDS) NOT INCLUDED IN THIS MAILING WILL FOLLOW UNDER SEP-

**HYDROCHLORIC ACID SOLUTIONS, IN TO K PAGE 01 OF 05

**HYDROCHLORIC ACID SOLUTIONS, IN TO N/1
**HYDROCHLORIC ACID SOLUTIONS, IN TO N/1
**HYDROCHLORIC ACID SOLUTIONS, IN TO N/1

MATERIAL SAFETY DATA SHEET

CHEMICAL DIVISION

1 REAGENT LANE

FAIR LAWN NJ 07410

(201) 796-7100

EMERGENCY CONTACTS
GASTON L. PILLORI
(201) 796-7100

DATE: 04/13/86 PO NBR: N/A ACCT: 818725-9

ACCT: 818725-99 INDEX: 25-8606-40098 CAT NO: SA601

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SUBSTANCE IDENTIFICATION

SUBSTANCE: **HYDROGRIGARIC ACID SOLUTIONS, IN TO N/100**

WATER WATER WATER TRADE NAMES/SYNONYMS: HYDROCHLORIC ACID 4%, WATER 96%; HYDROCHLORIC ACID 98%; HYDROCHLORIC ACID 0.68%, WATER >99%; HYDROCHLORIC ACID 0.068%, WATER >99%; HYDROCHLORIC ACID 0.0 ACID 2%, 0.04%,

CHEMICAL FAMILY: INORGANIC ACID

MOLECULAR FORMULA: H-CL N H20

CERCLA RATINGS (SCALE 0-3): NFPA RATINGS (SCALE 0-4): 1 HEALTH=3 HEALTH=3 FIRE=0 R REACTIVITY=0 PERSISTENCE=0

COMPONENTS AND CONTAMINANTS

PERCENT: 6 063-4 COMPONENT: HYDROCHLORIC ACID, CONCENTRATED CAS 7647-01-0

PERCENT: 96-99.932 COMPONENT: WATER

OTHER CONTAMINANTS: NONE

5 PPM OSHA CEILING; 5 PPM ACGIH CEILING

PHYSICAL DATA

DESCRIPTION: COLORLESS LIQUID BOILING POINT: 212 F

001)

S

THELTING VAPOR PRESSURE: 14 MMHG a POINT: 32 71 0 C 20 C (H20) SPECIFIC GRAVITY: EVAPORATION RATE: (ETHER=1) >1 1 . 0

**HYDROCHLORIC ACID SOLUTIONS, IN TO N
SOLUBILITY IN WATER: COMPLETE VAPOR DENSITY: 0.7 (H2O) VAPOR DENSITY: PAGE 02 OF 05

FIRE AND EXPLOSION

FIRE AND EXPLOSION HAZARD: HEGLIGIBLE FIRE AND EXPLOSION HAZARD WHEN EXPOSED TO HEAT OR FLAME

HOWEVER THE MORE CONCENTRATED TYPES TO EMIT FLAMMABLE HYDROGEN GAS. WILL REACT APPRECIABLY WITH COMMON METALS

FLASH POINT: WILL NOT BURN

FIREFIGHTING MEDIA: DRY CHEMICAL, CARBON DIOXIDE, WATER SPRAY (1984 EMERGENCY RESPONSE GUIDEBOOK, DOT P OR FOAM 5800.3).

FOR LARGER FIRES, ARGER FIRES, USE WATER SPRAY, I F0G OR ALCOHOL FOAM 5800.3).

FIREFIGHTING:

WEAR PERSONAL PROTECTIVE EQUIPMENT. MOVE CONTAINERS FROM FIRE AREA IF POSSIBLE. COOL CONTAINER EXPOSED TO FLAMES WITH WATER FROM SIDE UNTIL WELL AFTER FIRE IS OUT (1984 EMERGENCY RESPONSE GUIDEBOOK, DOT P 5800.3).

EXTINGUISH USING AGENTS INDICATED; IF LARGE AMOUNTS OF COMBUSTIBLE MATERIALS ARE INVOLVED, USE WATER SPRAY OR FOG IN FLOODING AMOUNTS. USE WATER SPRAY TO ABSORB CORROSIVE VAPORS. COOL CONTAINERS WITH FLOODING AMOUNTS OF WATER FROM AS FAR A DISTANCE AS POSSIBLE. AVOID BREATHING CORROSIVE VAPORS; KEEP UPWIND (BUREAU OF EXPLOSIVES, EMERGENCY HANDLING OF HAZARDOUS MATERIALS IN 1981).

TOXICITY

HYDROCHLORIC ACID: 1300 PPM/30 MINUTES
3124 FPM/1 HOUR INHALATION-RAT LC50;
900 MG/KG ORAL-RABBIT LD50; 40 MG/KG 81 MG/KG UNKNOWN-MAN LDLO; MUTAGENIC DATA (RTECS); CARCINOGEN STATUS: NONE HYDROGEN CHLORIDE VAPOR IS A SEVERE EYE, MUCOUS MEMBRANE, AND SKIN IRRITANT INHALATION-HUMAN LCLO; 1108 PPM/1 HOUR INHALATION-MOUSE LC50; INTRAPERITONEAL-MOUSE LD50

HEALTH EFFECTS AND FIRST AID

CORROSIVE/TOXIC. RROSIVE/TOXIC. 100 PPM (HCL) IMMEDIATELY DANGEROUS TO LIFE OR HEALTH. ACUTE EXPOSURE- EXPOSURE TO GAS OR FUMES CAUSES IMMEDIATE COUGHING, BURNING OF THE THROAT OR NOSE, CHOKING, DIZZINESS, WEAKNESS AND DIFFICULTY SWALLOWING. EXPOSURE ABOVE 5 PPM ARE FOLLOWED BY INFLAMMATION AND OCCASIONAL ULCERATION OF THE NOSE, THROAT OR LARYNX; LARYNGITIS, BRONCHITIS, PREUMONIA, HEADACHE, PALPITATIONS, DENTAL EROSION, OR NASAL SEPTUM PERFORATION. CONCENTRATIONS ABOVE 50 PPM MAY BE FOLLOWED BY BLEEDING OF THE NOSE AND GUTS. FOLLOWING A 6-8 HOUR LATENCY PERIOD, LARYNGEAL SPASM OR PULMONARY EDEMA WITH TIGHTNESS IN THE CHEST, AIR HUNGER, DIZZINESS, FROTHY SPUTUM AND CYANOSIS MAY OCCUR. SHORTNESS OF BREATH AND EXPECTORATION OF BLOOD MAY OCCUR FOR SEVERAL WEEKS, FOLLOWING A SINGLE EXPOSURE. PYLORIC

ASPHYXIATION, **HYDROCHLORIC GASTRIC HEMORRHAGE, ACID SOLUTIONS, INFECTION, CYAN CYANOSIS AND OI NI ż PAGE DEATH. 03 0 05

RONIC EXPOSURE- CAUSES EROSION OF TEETH FOLLOWED BY JAW NECROSIS, BRONCHIAL IRRITATION WITH CHRONIC COUGH, FREQUENT ATTACKS OF BRONCHIAL BRONCHIAL FNEUMONIA, SKIN TENDERNESS, GASTROINTESTINAL DISTURBANCES OR MUCOUS MEMBER OF THE UPPER RESPIRATORY BRANE IRRITATION WHICH MAY MIMIC VIRAL INFECTION OF THE UPPER RESPIRATORY TRACT CHARACTERIZED BY FEVER AND MUSCLE TENDERNESS. SEE MUTAGENIC DATA REFERENCE IN TOXICITY SECTION.

FIRST AID- REMOVE FROM EXPOSURE AREA TO FRESH AIR IMMEDIATELY. IF BREATHING HAS STOPPED, FERFORM ARTIFICIAL RESPIRATION. KEEP AFFECTED PERSON WARM AND AT REST. GET MEDICAL ATTENTION.

CORROSIVE.

ACUTE EXPOSURE- CAUSES SEVERE PAIN, YELLOWISH-BROWN STAINS, BURNS ON SEVERE EXPOSURE, SHOCK SYMPTOMS OF RAPID PULSE, SWEATING AND MAY OCCUR.

CHRONIC EXPOSURE- FROLONGED VAPOR CONTACT MAY CAUSE DERMATITIS

FIRST AID- DIRECT A STREAM OF WATER UNDER CONTAMINATED CLOTHING AND SHOES WHILE REMOVING. WASH AFFECTED AREA WITH SOAP OR MILD DETERGENT AND LARGE AMOUNTS OF WATER UNTIL NO EVIDENCE OF CHEMICAL REMAINS (APPROXIMATELY DRESSING. 15-20 MINUTES). BANDAGE SECURELY, BUT NOT IN CASE OF CHEMICAL BURNS, COVER AREA WITH STERILE, DRY GET MEDICAL ATTENTION

CORROSIVE.

ACUTE EXPOSURE- MAY CAUSE FAIN, TEARING, REDNESS, OR BLURRED VISION. EY CONTACT CAUSES IRREVERSIBLE CORNEAL INJURY IF NOT IMMEDIATELY WASHED. CONTACT WITH HIGH CONCENTRATIONS CAUSES CONJUNCTIVAL EDEMA, CORNEAL DESTRUCTION OR LOSS OF VISION.

CHRONIC EXPOSURE- PROLONGED CONTACT MAY CAUSE RESULT IN IMPAIRMENT OR LOSS OF NOISIA REDNESS AND IRRITATION AND MAY

FIRST AID- WASH EYES IMMEDIATELY WITH LARGE AMOUNTS OF WATER, OCCASIC LIFTING UPPER AND LOWER LIDS, UNTIL NO EVIDENCE OF CHEMICAL REMAINS (APPROX. 15-20 MIN.). IN PRESENCE OF BURNS, APPLY STERILE BANDAGES GET MEDICAL ATTENTION. REMAINS OCCASIONALLY

CORROSIVE/TOXIC

ACUTE EXPOSURE- CAUSES BURNS OF MOUTH, ESOPHAGUS, AND STOMACH WITH CONSEQUENT PAIN, NAUSEA, SALIVATION, VOMITING, CHILLS, SHOCK AND THIRST. CAUSES ULCERATION OF ALL MEMBRANES AND TISSUES WHICH THE ACID CONTACYS. ASPHYXIA OR NEPHRITIS MAY OCCUR. AFTER INITIAL RECOVERY, FEVER MAY INDICATE PERFORATION OF THE ESOPHAGUS OR STOMACH. IN SEVERE CASES, CIRCULATORY COLLAPSE OCCURS WHICH, IF NOT CORRECTED, MAY LEAD TO RENAL, LIVER OR HEART FAILURE.

FIRST AID- IF VICTIM IS IMMEDIATELY TO DILUTE OUNCE (30 ML) OF MILK CONSCIOUS, GIVE HIM LARGE QUANTITIES OF WATER THE ACID. DO NOT INDUCE VOMITING. GIVE PATIENT OF MAGNESIA. GET MEDICAL ATTENTION IMMEDIATELY.

REACTIVITY

REACTIVITY: STABLE UNDER NORMAL PRESSURES 10 BOILING POINT, 100 0

INCOMPATIBILITIES:

METALS AND WATER-REACTIVE MATERIALS EXAMPLES FOLLOW:

ALCOHOLIC HYDROGEN CYANIDE: TETRASELENIUM TETRANITRIDE: EXPLOSIVE REACTION.

ODIUM: EXPLOSIVE REACTION

OTASSIUM PERMANGAHATE: EXPLOSIVE REACTION.

ULFURIC ACID: EXPLOSIVE REACTION. ERCHLORIC ACID: VIOLENT REACTION.

CETIC ANHYDRIDE, CLOSED CONTAINER. ETIC ANHYDRIDE, 2-AMINOETHANOL, 28% AMMONIA, DIAMINE, ETHYLEHEIMINE, OLEUM, PROPIOLACTONE CHLOROSULFONIC ACID, ETI (BETA-), PROPYLENE OXID ETHYLENE-

LUORINE: IGHITES.

IUM CARBIDE: IGHITES.

CESIUM ACETYLIDE: IGHITES. RUBIDIUM CARBIDE: IGHITES.

RUBIDIUM ACETYLIDE:

JRANIUM DICARBIDE: IGNITES.

JYYGEN + PLATINUM: IGNITES.

LITHIUM SILICIDE: INCANDESCENCE.

ALUMINUM: VIGOROUS EXOTHERMIC REACTION.

ALCIUM PHOSPHIDE: VIGOROUS EXOTHERMIC REACTION RANIUM PHOSPHIDE: PRODUCES EXPLOSIVE OR FLAMMA

RANIUM PHOSPHIDE: PRODUCES EXPLOSIVE OR FLAMMABLE PHOSPHINE. EXPLOSIVE OR FLAMMABLE

DOWICIL 100: DECOMPOSES DOWICIL 100. SULPHIDES: REACTS TO FORM CHLORIDES AND HYDROGEN SULPHIDE TRICHLOROMETHYL PERCHLORATE.
METALS: PRODUCES EXPLOSIVE OR FLAMMABLE
MAGNESIUM BORIDE: PRODUCES EXPLOSIVE OR HYDROGEN GAS. FLAMMABLE PRODUCT.

RELEASES HYDROGEN.

DECOMPOSITION: HEATING INCREASES

THE RATE 유 RELEASE 0F CORROSIVE HYDROGEN CHLORIDE

POLYMERIZATION: NOT OCCUR

CONDITIONS TO AVOID

FLAMMABLE, POISONOUS GASES MAY ACCUMULATE IN TANKS AND HOPPER CARS. AVOID HEATING TO THE POINT WHERE AN UNACCEPTABLE ANOUNT OF HYDROGEN CHLORIDE IS EVOLVED. AVOID CONTACT WITH OR STORAGE WITH INCOMPATIBLE MATERIALS, INCLUITHOSE LISTED IN THE REACTIVITY SECTION. INCLUDING

SPILL AND LEAK PROCEDURES

PROVIDE VENTILATION.

WEAR PERSONAL PROTECTIVE EQUIPMENT. COVER WITH SODA ASH.

UP AND PLACE 'CORROSIVE'. **HYDROCHLORIC ACID SOLUTIONS, IN TO N IN A SUITABLE NONMETALLIC CONTAINER, CLOSE KEEP OUT OF SEWERS AND WATER SOURCES. PAGE 05 9 F 05

PROTECTIVE EQUIPMENT

VENTILATION: FROVIDE LOCAL REQUIREMENTS. EXHAUST VENTILATION SYSTEM TO NEET PERMISSABLE EXPOSURE EQUIPMENT MUST BE RESISTANT TO CORROSION BY ACIDS.

RESPIRATOR:

EXPOSURE LIMIT TO 50 PPM-

CHEMICAL CARTRIDGE RESPIRATOR WAITH ACID SUPPLIED-AIR RESPIRATOR SELF-CONTAINED BREATHING APPARATUS. GAS CARTRIDGE(S).

>50 PPM, INCLUDING THE IDLH LEVEL, 100 PPM
SELF-CONTAINED BREATHING APPARATUS WITH A
OPERATED IN PRESSURE-DEMAND OR OTHER POS
OR USE EQUIVALENT RESPIRATOR. POSITIVE FULL FACEPIECE SITIVE PRESSURE

MODE

FIREFIGHTING- SELF-CONTAINED BREATHING APPARATUS WITH A FULL FACEPIECE OPERATED IN PRESSURE-DEMAND OR OTHER POSITIVE PRESSURE MODE.

CLOTHING: EMPLOYEE MUST WEAR APPROPRIATE PROTECTIVE CLOTHING AND ANY FOSSIBILITY OF SKIN CONTACT WITH THIS SOLUTION. EQUIPMENT J

EMPLOYEE MUST WEAR APPROPRIATE SOLUTION. PREFERRED MATERIALS: LIMITED DATA) PROTECTIVE GLOVES TO POLYETHYLENE, VITON, PREVENT CONTACT WITH THIS AND SARANEX (BASED ON

EYE PROTECTION:
EMPLOYEE MUST WEAR
ANY POSSIBILITY OF WHEN WORKING WITH CHEMICALS. SPLASH-PROOF SAFETY GOGGLES CONTACT WITH THIS SOLUTION. AND A FACESHIELD TO DO NOT WEAR CONTACT PREVENT LENSES

WHERE THERE IS THIS SOLUTION, IMMEDIATE WORK ANY POSSIBILITY THAT AN EMPLOYEE'S EYES MAY BE EXPOSED THE EMPLOYER SHALL PROVIDE AN EYE-WASH FOUNTAIN WITHIN AREA FOR EMERGENCY USE.

AUTHORIZED CREATION DATE:] 10/29/85 - ALLIED FISHER SCIENTIFIC REVISION DATE:

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